H-4/28/22

Roll No.

IV Semester Examination, 2022

M.Sc.

PHYSICS

Paper I

(Atomic and Molecular Physics)

Time: 3 Hours]

Max. Marks: 80

Note: All questions are compulsory. Question Paper comprises of 3 sections. Section **A** is objective type/multiple choice questions with no internal choice. Section **B** is short answer type with internal choice. Section \mathbf{C} is long answer type with internal choice.

SECTIONA

 $1 \times 8 = 8$

(Objective Type Questions)

- 1. The total energy of the electron in 2^{nd} orbit of hydrogen atom is
- **2.** Write the quantum condition for the elliptic orbits.
- **3.** What is Larmour precession?
- 4. Drow diagram for Anomalous Zeeman Pattern.
- **5.** Define linear top molecule.

- **6.** Pure rotational molecular spectra observed in the resion of the electromagnetic spectrum.
- 7. For R-branch and P-branch which is correct.

(a)
$$\Delta J = +1$$
, $\Delta J = +1$

(b)
$$\Delta J = +1$$
, $\Delta J = -1$

(c)
$$\Delta J = -1$$
, $\Delta J = +1$

(d)
$$\Delta J = -1$$
, $\Delta J = -1$

8. Write formation off electronic spectra.

SECTION B

 $6 \times 4 = 24$

(Short Answer Type Questions)

Note: Attempt *one* question from each unit.

Unit-I

1. Write Bohr's postulates and determine the total energy of electron.

Or

Describe vector atom model.

Unit-II

2. Explain orbital magnetic dipole moment and define Bohr magneton.

Or

Write a note on nuclear magnetic resonance.

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Unit-III

3. Explain symmetric top assymenetric top and spherical top molecules with example.

Or

Determine the equation of energies for harmonic oscillater.

Unit-IV

4. Explain rotational, vibrational and electronic spectra of molecules.

Or

Write note on Franck Condon principle.

SECTION C

 $12 \times 4 = 48$

(Long Answer Type Questions)

Note: Attempt *one* question from each unit.

Unit-I

1. Prove that the number of terms for *pd* configuration is the same for L-S coupling as that for *j-j* coupling.

Or

Describe the general features of spectra of alkalilike atoms and explained it.

Unit-II

2. What is Paschen-Back effect? Explain the splitting of sodium lines with the help of it.

Or

Describe normal and anomalous Zeeman's effect. How they are explained by quantum theory?

Unit-III

3. Derive the expression for the energy of a rigid-rotation model of a diatomic molecules and explain the rotational spectrum.

Or

Discuss molecule as Anharmonic oscillator and calculate vibrational frequency and force constant for anharmonic oscillator.

Unit-IV

4. Explain P, Q and R branches in vibration-spectra of vibrating rotator.

Or

- (a) Explain Born-oppenheimer approximation.
- (b) Explain formation of electronic spectra.
