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Roll No.....

# M.Sc. I Semester Examination, April-2021

## CHEMISTRY

#### Paper IV

(Basics of Spectroscopy)

#### Time : 3 Hours ]

#### [Maximum Marks : 80

Note : All questions are compulsory. Question Paper comprises of 3 sections. Section A is objective type/Multiple Choice questions with no internal choice. Section B is short answer type with internal choice. Section C is long answer type with internal choice.

# SECTION 'A' 1×8=8

(Objective Type Questions)

Choose the correct answer :

- 1. Following pairs of molecules do not show IR-Ratational spectrum :
  - (a)  $CO_2$  and  $NO_2$  (b)  $O_2$  and  $H_2$
  - (c)  $N_2$  and  $D_2O$  (d)  $Cl_2O$  and  $N_2$
- 2. Which one of the following are microwave active :
  - (a)  $O_2$  (b)  $CO_2$
  - (c)  $CH_3Cl$  (d)  $N_2O$

Write the basic principles of photoelectron spectroscopy.

4. Which one of the following stretching vibration shown below is IR-inactive ?

	$\leftarrow \leftarrow$
S=C=S	S=C=S
$\leftarrow \rightarrow$	
(I)	(II)

Or

What is mutual exclusion principles ?

*SECTION 'C'* **12×4=48** 

(Long Answer Type Questions) Note : Answer the following questions in 500 words.

1. Describe the stark effect.

#### Or

Discuss rotational, vibrational and electronic energy level with diagram.

2. Explain the orbital angular momenta in  $P^2$ ,  $P^3$  and  $d^2$  configuration.

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- 8. Total number of vibrations in allyl bromide CH<sub>2</sub> = CHCH<sub>2</sub>Br are :
  - (a) 18 (b) 21
  - (c) 14 (d) 16

*SECTION 'B'* **6×4=24** 

(Short Answer Type Questions)

Note : Answer the following questions in 250 words.

1. Write note on intensity of spectral lines of rotational spectrum.

### Or

Explain the selection rule for transition between the rotational energy levels.

2. Calculate the  $\lambda_{max}$  in UV-spectrum of 2, 4-hexadiene



Or

Explain Lambert-Beer's Law.

3. Discuss the charge transfer spectra.

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- 3. The ground term for  $3d^9$  configuration is :
  - (a) 3P (b)  $^{2}D$
  - (c)  ${}^{1}D$  (d)  ${}^{1}S$
- 4. In alkali metal spectra energy component level (excluding S level) is :
  - (a) Doublet (b) Singlet
  - (c) Triplet (d) Quartet
- 5. Sources of irradiation in ESCA is :
  - (a) Halogen light (b) Sodium vapour lamp
  - (c) Monochromatic X-rays
  - (d)  $\gamma$ -rays
- 6. Fermi level is :
  - (a) d-d transition (b)  $M \rightarrow L$
  - (c) D<sup>2/5</sup> state (d) Zero potential energy
- 7. No rotational Raman effect is observed for :
  - (a) Spherical top molecules
  - (b) Symmetric top molecules
  - (c) Linear molecules
  - (d) Asymmetric top molecules

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#### Or

Explain the absorption bands in UV-spectrum.

3. Discuss the electronic spectra of polyatomic molecules.

#### Or

Write the basic ideas of Auger-effect.

4. Discuss the fundamental bands and overtone.

#### Or

Write note on Coherent Antistokes Raman Spectroscopy (CARS).

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