G-2/214/21

Roll No.

M.Sc. II Semester Examination, 2021 **CHEMISTRY**

Paper II

(Organic Chemistry)

Time: 3 Hours]

Max. Marks: 80

Note: All questions are compulsory. Question Paper comprises of 3 sections. Section A is objective type/multiple choice questions with no internal choice. Section B is short answer type with internal choice. Section C is long answer type with internal choice.

SECTIONA

 $1 \times 8 = 8$

(Objective Type Questions)

Choose the correct answer:

- 1. The reaction in which silver salts of carboxylic acids react with a halogen to give an organic halide is:
 - (a) Hunsdieker Reaction
 - (b) Sandmeyer Reaction
 - (c) Aldol Reaction (d) Benzoin reaction
- **2.** The product of the reaction

$$\begin{array}{c|c} \operatorname{CH_3} & & \\ & | & \\ \operatorname{C_2H_5Na} + \operatorname{H_3C} - \operatorname{C} - \operatorname{CH_3} & \xrightarrow{\operatorname{C_2H_5OH}} & \operatorname{is} : \\ & | & \\ \operatorname{Br} & & \operatorname{P.T.O.} \end{array}$$

(a)
$$H_3C-C-CH_3$$
 (b) $CH_3-C=CH_2$ | H

(c)
$$H_2C = C - CH_2Br$$
 (d) $C_2H_5 - C - CH_3$ | CH_3 | H

- **3.** Propene on hydroboration gives :
 - (a) Propan-1-ol
- (b) Propan-2-ol
- (c) Propanone
- (d) None of these
- **4.** Michael addition reaction involves :
 - (a) Carbocation
- (b) Free radical
- (c) Carbanion
- (d) None of these
- **5.** Lithium aluminium hydride reduction of 4-t-butyl cyclohexanone gives which alcohol?
 - (a) Axial alcohol
- (b) Equatorial alcohol
- (c) Both (a) and (b) (d) None of these
- **6.** Wittig reaction uses:
 - (a) Carbonyl compounds with ylides
 - (b) Alcohol with ylides

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- (c) Phosphate esters
- (d) None of these
- **7.** A 1, 3 migration of carbon can take place thermally with of configuration.
- **8.** Pericyclic reaction are concerted, unaffected by catalysts or solvent and have transition state.

SECTION B

 $4 \times 6 = 24$

(Short Answer Type Questions)

Note: Answer the following questions in 250 words.

Unit-I

1. Explain Neighbouring Group Assistance in detail.

Or

Explain the mechanism of E_1 and E_2 reactions.

Unit-II

2. Write explanatory notes on Mechanistic and stereo chemical aspects of addition reaction involving electrophiles.

Or

Explain Hydrogenation of double and trible bonds with suitable examples.

Unit-III

3. Complete the following reactions :

(a)
$$ClCH_2CH_3 \xrightarrow{\text{(i) } Ph_3P}$$

$$Ph_3P^+$$
— $C\overline{H}CH_3$ —Base—.....

(b)
$$R - C - OR' \xrightarrow{H_3O^+}$$

(c)
$$2C_6H_5CHO \xrightarrow{KCN}$$

Or

Explain the mechanism of Hydrolysis of esters and amides.

Unit-IV

4. Explain cycloaddition antarafacial and suprafacial addition.

Or

Explain FMO approach for cycloaddition reaction.

SECTION C

$12 \times 4 = 48$

(Long Answer Type Questions)

Note: Answer the following questions in 500 words.

Unit-I

- 1. Explain:
 - (i) Auto oxidation coupling of alkynes.
 - (ii) Arylation of Aromatic compounds by diazonium salts.

Or

Explain:

- (i) The spectrum of E_1 , E_2 and E_1CB mechanism.
- (ii) Mechanism and orientation in Pyrolytic elimination.

Unit-II

2. Discuss the addition reaction to cyclopropane ring and hydrogenation of aromatic rings.

Or

Explain in detail:

- (i) Michael Reaction,
- (ii) Sharpless Asymmetric Epoxidation.

Unit-III

3. How an aldol reaction can be made regioselective, diastereo selective and enantio

selective? Explain taking suitable compounds and only the reagents and conditions in each case.

Or

Give the mechanism and applications of the following reactions :

- (i) Perkin Reaction,
- (ii) Mannich Reaction,
- (iii) Stobbe Reaction.

Unit-IV

- **4.** Write explanatory notes on :
 - (i) Fluxional Tautomerism,
 - (ii) Sigmatropic Rearrangements,
 - (iii) Conrotatory and Disrotatory motions.

Or

Explain taking one example as to low FMO method, PMO method and correlation diagram can be used for analysing a Pericyclic reaction.

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